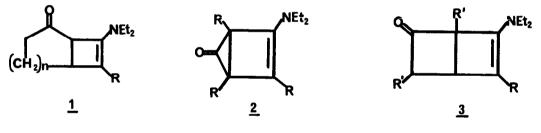
UNEXPECTED REARRANGEMENT DURING THE CYCLOADDITION OF YNAMINES WITH CYCLOBUTENONES. Jacqueline Ficini, Samir Falou, Jean d'Angelo. Laboratoire de Chimie Organique de Synthèse Equipe de Recherche Associée au C.N.R.S. Université Pierre et Marie Curie. 8, rue Cuvier - 75005 PARIS.

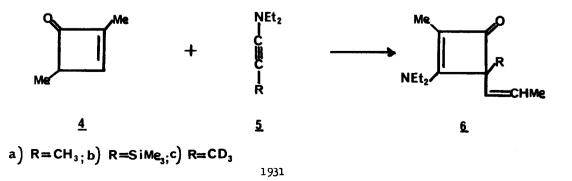
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The cycloadducts <u>1</u> obtained from ynamines ¹ and cyclopentenones ² or cyclohexenones ³ are, as expected, thermally stable ^{4a}, whereas the very strained cycloadduct <u>2</u> obtained from diphenyl-cyclopropenone ⁵ undergoes, in situ, a rearrangement which does not follow the selection rules ⁴.

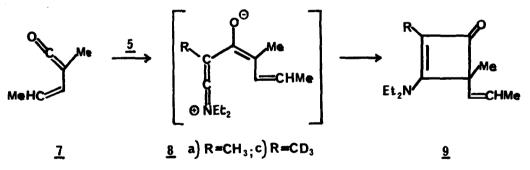


n=1 or 2

We now report the first results concerning the reaction of ynamines with cyclobutenones. These unexpected results which deal with 2,4 dimethylcyclobutenone $\frac{4}{2}^{6}$ and the ynamines $\frac{5}{2}^{-1}$ show that, in all cases, the product is not an amino bicyclo (2-2-0) hexenone of type $\frac{3}{2}$, but an amino-cyclobutenone $\frac{6}{2}$ which arises from a very unusual rearrangement.

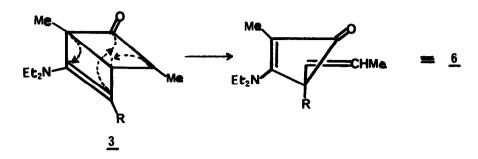


The structure proposed for amino-cyclobutenones <u>6</u>, is in agreement with the results of elementary analysis, Mass Spectra, IR and NMR spectroscopic data. This structure has been established by comparison with samples obtained <u>via</u> the reaction ¹⁰ of ynamines <u>5a</u> and <u>5c</u> (0.5 equivalent) with ketene <u>7</u> ¹¹. When R is a methyl group (<u>5a</u> : $R = CH_3$), the cycloadduct <u>9a</u> is identical (IR, NMR, Mass Spectrum) with amino-cyclobutenone <u>6a</u>, but when R is deuteromethyl group (<u>5c</u> : $R = CD_3$) the cycloadduct <u>9c</u> : [B.p. 80°-90° (0.01 mm) ; $M^+ = 210$; 75 % yield ; IR (neat) = 1745, 1600-1570 (broad) cm⁻¹ ; NMR ¹² (CCl₄) 1.21, 1.29 and 1.33 (t,t and s,9H), 1.76 (d,d,3H), 3.1-3.5 (m,4H), 5.49 (m,2H)] is by contrast, very different (NMR) from its regioisomer <u>6c</u>.

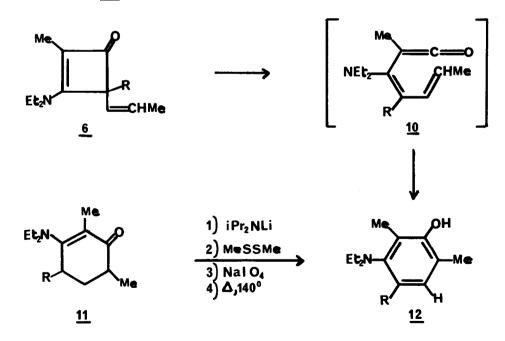


These results suggest that the initial step of the cycloaddition of

ynamines with cyclobutenones is an amino bicyclo (2-2-0) hexenone such as <u>3</u>. This bicyclic adduct <u>3</u> rearranges <u>in situ</u>, to give regiospecifically the cyclobutenone <u>6</u> by a concerted process which involves one π bond and two σ bonds $(\pi^2 + \sigma^2 + \sigma^2)$. For such a process to be Woodward-Hoffmann allowed the number of suprafacial (4q + 2) elements must be odd ${}^{4b}, c$. This implies that, in particular, this rearrangement might be $(\pi_s^2 + \sigma_s^2 + \sigma_s^2)^{-13}$. The aromatization of the bicyclic adduct <u>3</u> does not occur, at this temperature, for this disrotatory process which would have given phenols <u>12</u>, is thermally forbidden 4a .



By contrast, the aromatization of amino-cyclobutenone <u>6</u> is very easy and leads to phenols <u>12</u>. For instance, the phenol <u>12a</u>: B.p. 100°-110° (0.01 mm); IR (neat): 3600-3100, 1580, 805 cm⁻¹; NMR (CCl₄): 0.95 (t,6H), 2.1 (s,9H), 3.05 (q,4H), 6.6 (s,1H) is obtained by heating <u>6a</u> above 140°. In this case, one can assume that the easy thermal opening of the cyclobutenone ring ¹⁴ which leads to ketene <u>10</u> is followed by the cyclization ⁴ of the triene <u>10</u> ¹⁵. One notices that when R is a trimethylsilyl group, it disappears during the aromatization. The phenol <u>12d</u>: [B.p. 80°-100° (0.05 mm); IR (neat): 3600-3100, 1585, 800 cm⁻¹; NMR (CCl₄): 0.85 (t,6H), 2.1 (s,6H), 2.85 (q,4H), 6.5 (d,J = 8 Hz,1H), 6.8 (d,J = 8 Hz,1H)] obtained in 60 % yield from <u>6b</u> ¹⁵ is, indeed, identical with an authentic sample prepared from <u>11d</u> ¹⁶, according to the pathway indicated, cf ref. 17.



a) $R = Me_{1} b$ $R = Si(Me)_{3} c$ $R = CD_{3} d$ R = H

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- 11) 7 is prepared according to H. MAYR Angew. Chem. Int. Ed. 14, 500 (1975), from 4 by refluxing in cyclohexane : after 1 h 30, 80 % (by NMR) of 4 is converted into 7.
- 12) NMR spectra of all compounds described in this paper are taken on a Varian T 60 Apparatus, except <u>9c</u> and <u>6c</u> taken on Cameca 250 MHz Apparatus for which we thank Dr. LALLEMAND -Ecole Normale Supérieure, Paris.
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